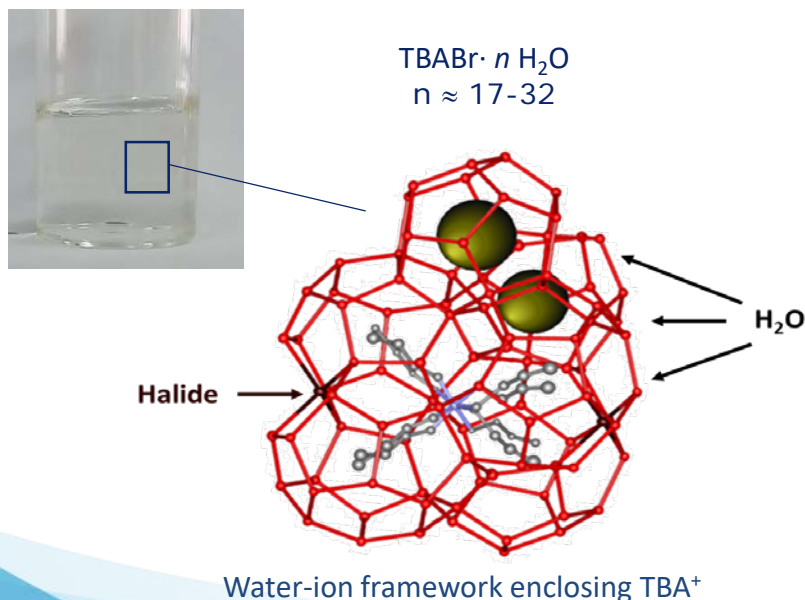


In the Liquid Structure:

- Both **water molecules** and **anions** (halide) are tightly held in the H-bonded network structure with a reduced reactivity.
- **Small empty cages** can capture small-size non polar molecules (host-nanocages).



Key Features (Energy Storage and Conversion):

- Do not restrict electrode selection
- Widening the potential stability window
- Energy density enhancement
- Alternative to highly concentrated “water-in-salt”
- Stable under atmospheric conditions
- Highly incompressible
- Properties could be optimized for different applications
- Could be extended to other pnictogen alkyl salts

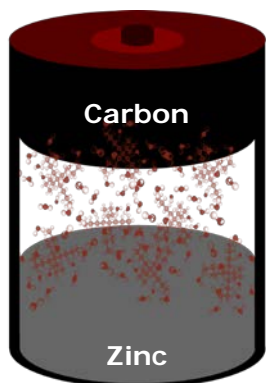


By adjusting the concentration of tetrabutyl ammonium and phosphonium salts in water ($\approx 1.5\text{--}2.0\text{ m}$), hydrophobic solvation triggers the formation of a unique, highly incompressible supramolecular liquid, with a dynamic structure similar to clathrates, involving essentially all H₂O molecules of the solvent

[Angew. Chem. Int. Ed., 2021, 60, 7540](#)

Internal Proof of Concept: Supramolecular Zn-Bromine Battery

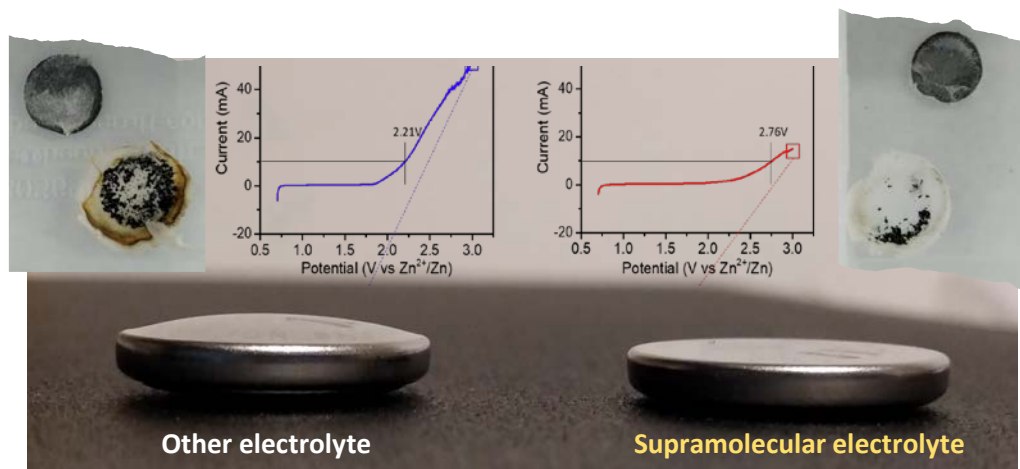
Can we confine Br₂ to enhance coulombic efficiency avoiding side reactions?



Cathode

Supramolecular
Electrolyte

Anode



It improves the stability of a zinc-bromine battery, hindering the electrolyte decomposition, and increase the capacity due to the reversibility of Br⁻/Br₂

Advantages of a supramolecular aqueous Zn-Bromine battery

- Metal-free electrode
- Large-scale compatible
- Cheap and eco-friendly
- No dendrite formation
- No water decomposition
- No cathodic intercalation of Zn²⁺
- Use the Br⁻/Br₂ redox process